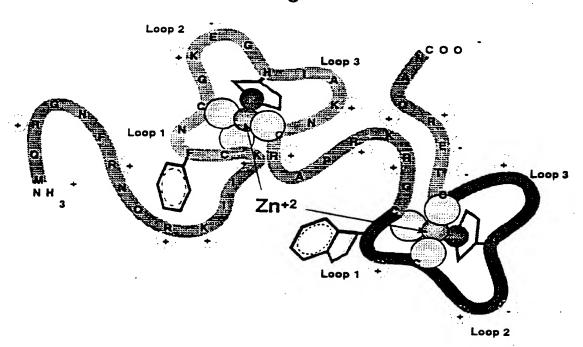


Figure 2

HIV-1_{MN} Nucleocapsid Protein

	first array linker		second array .	
MORGNERNORKIIKCEN	CGKEGHIAKNO	RAPRKRGC	WKCGKEGHQI	∰⊝ ⊝⊕ MKDCTERQAN
. 0	o & c		8° 8	0
Total Residues55 Basic Residues15 Acid Residues4			Moiecular Weight	6451.5
Net Charge +11 IEP 10.77			280nm Molar	

First Zinc Finger

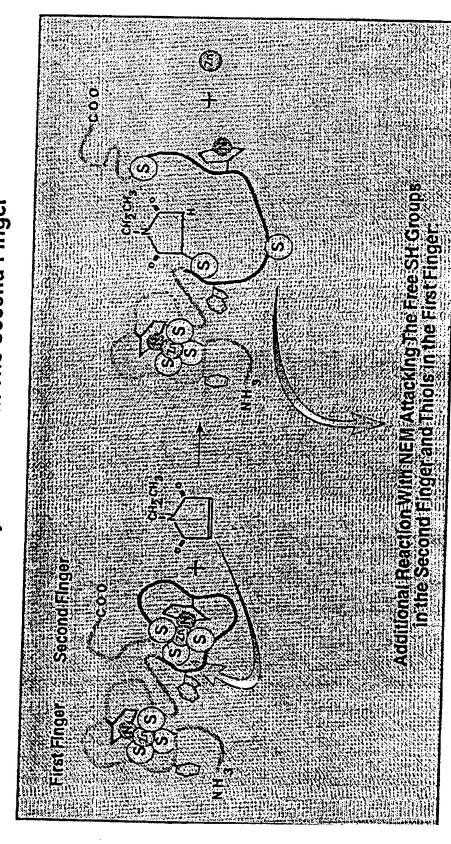


Second Zinc Finger

Figure 3

Figure 4

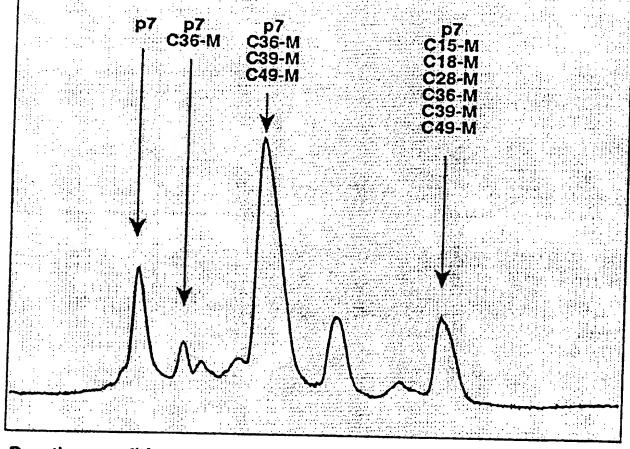
The Initial Reaction With NEM Modifies The First Cys Residue In The Second Finger



This is an example showing how the procedures have been used to investigate the reaction pathway and to determine the most reactive thiol in the NC protein. By reacting p7NC with limiting amounts of NEM and analyzing as in Fig. 6, it was determined that the first cysteine in the second zinc finger reacts fastest with the reagent.

Analysis of Reaction Products by HPLC

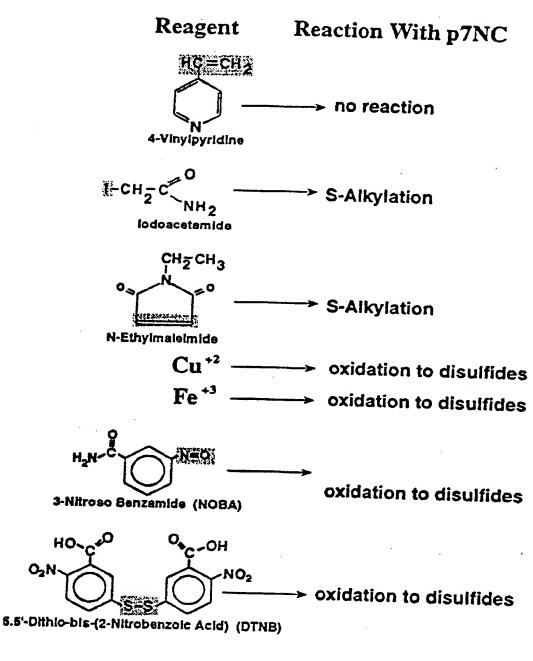
The positions of alkylated Cys residues were determined by sequence analysis of separated proteins and are indicated by the notation C#-M.



Reaction conditions: 62 mM p7NC + 744 mm NEM; pH 7.0, 60min. at RT.

Separation was accomplished by reversed phase HPLC using a C-18 μ -Bondapak (3.9 x 300 mm) column (Waters, Inc). Proteins were eluted at a flow rate of 1.0 ml/min. with gradients of acetonitrile (0-17, 20 min. 17-25, 120 min.) at pH 2.0 (0.05% trifluoroacetic acid). Proteins were detected by UV absorption at 206 nm.

REACTIONS OF HIV-1 NC RETROVIRAL CCHC ZINC FINGERS





The reactive funtional groups are shaded



Figure 7

Functional Groups Which React With Retroviral Zinc Fingers

disulfides

R—S—S—R

nitroso compounds

R-N=O

maleimides

 α -halogenated ketones

phenylhydrazids

Nitric Oxide and Derivitives

NO

cupric ions and complexes

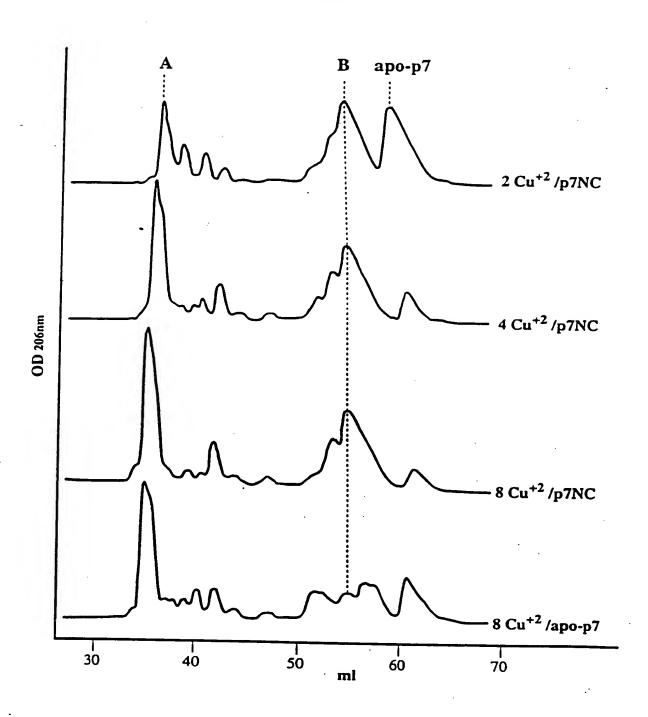
Cu⁺²

ferric ions and complexes

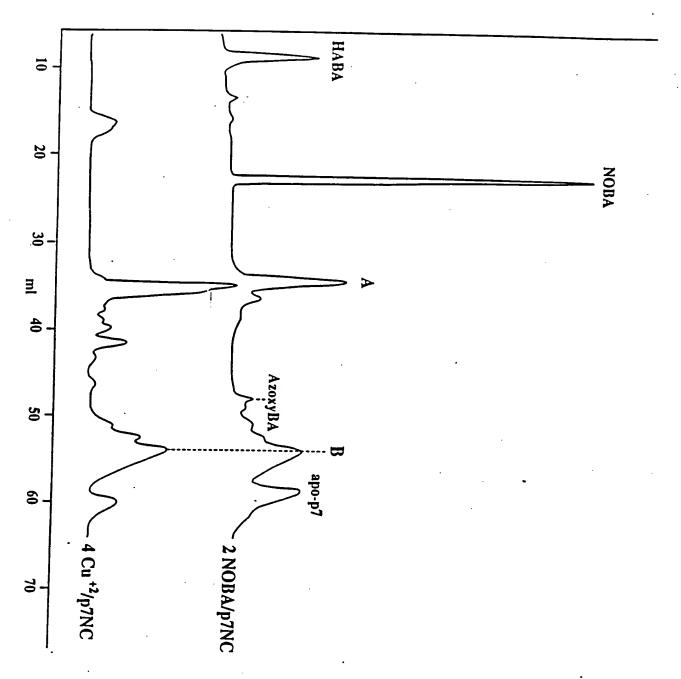
Fe⁺³

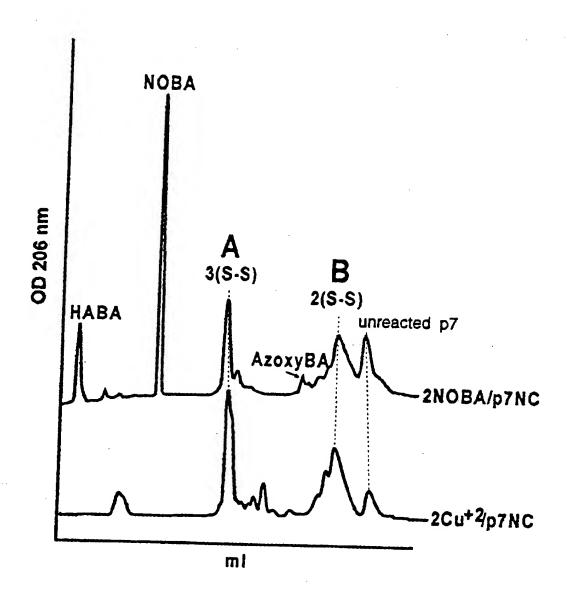
wherein R is any atom or molecule, and X is selected from the group consisting of F, I, Br and Cl.

FIGURE 8



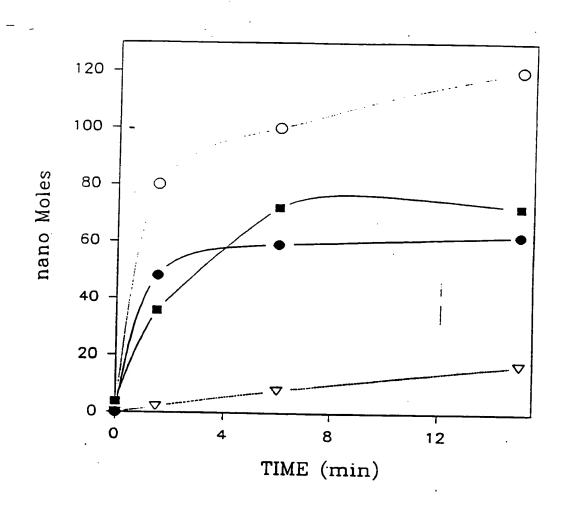
OD 206 nm

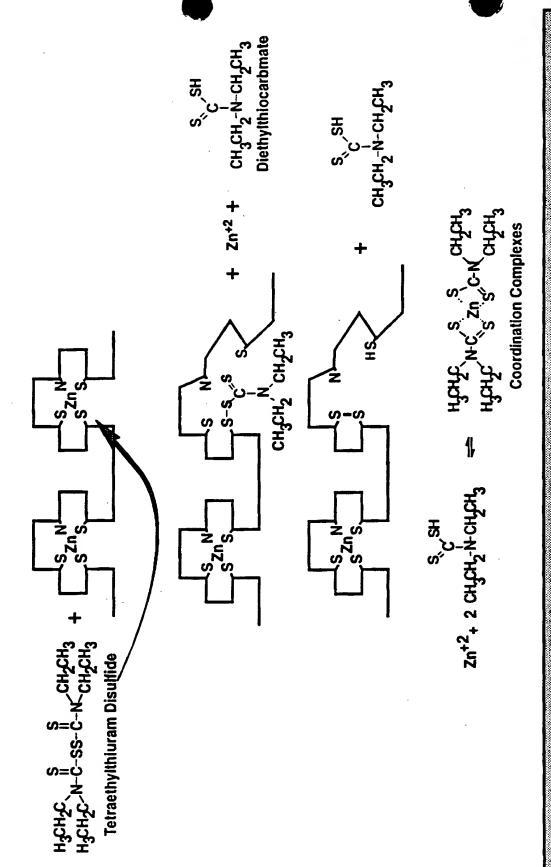




HPLC Chromatograms of NOBA and Cupric Oxidation Products of p7NC

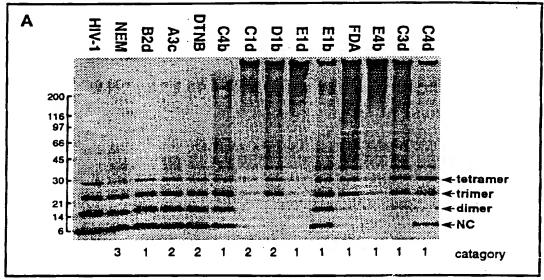
FIGURE 11

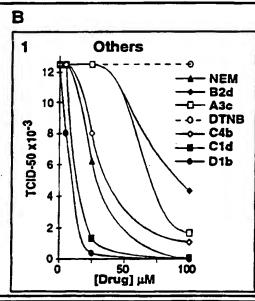


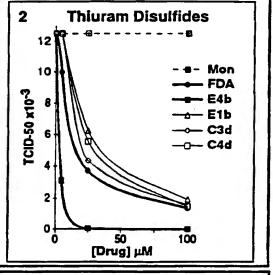


3 Tetraethylthluram Disulfide + p7NC -> Oxidized p7 (3 S-S) + 6 Diethylthlocarbamate +

4 Diethylthiocarbamate + 2 Zn+2 - - 2 Coordination Complexes







С	Compounds	Oral Toxicity (Mouse) LD-50 (g/kg)
FDA E4b E1b C4d C3d E1d D1b C1b C4b A3c B2d	(Tetraethylthiuram disulfide) (Dicyclopentamethylenethiuram disulfide) (Tetramethylthiuram disulfide) (Tetrabutylthiuram disulfide) (Tetraisopropylthiuram disulfide (2,2'-Dithiobis(Pyridine N-Oxide) (Aldrithiol-2) (2,2' Dithiobis(benzonitrile) (Foramidine disulfide) (Benzoly Disulfide) (4-(dimethylamino)phenyl disulfide)	1.98 2.87 2.35 1.35

Medical Use and Chemistry of Thiurams

Synthesis

General Reactions

HPLC Analysis Of p7NC Reactions With Imuthiol and Disulfiram

